STUDY OF POLYVINYL ACETATE FILMS TRANSFORMATION BY IR REFLECTANCE SPECTROSCOPY

Simion Jitian University "Politehnica" Timisoara, Faculty Engineering Hunedoara, Revolutiei, 5, Hunedoara, 331128, Romania

ABSTRACT

The transmitance values measured in IR reflection-absorption (RA) spectra can be used to determine the optical constants of dielectric films laid on solid substrates.

To obtain the optical constants of polymers films laid on steel we used dispersion analysis. In this case, the optical constants are obtained from IR spectrum recorded at a single incidence angle. Using dispersion analysis offers the advantage of processing a large volume of data.

Keywords:

Reflection-absorption, optical constants, IR spectra, dispersion analysis

1. INTRODUCTION

The study of reflection spectra of surface films deposited on solid media allow to determine the thickness and optical constants of these: the refractive index \mathbf{n} and absorption index \mathbf{k} .

The reflection of radiation on solid surfaces or transmission is expressed by its complex reflection $\tilde{r} = |\tilde{r}| \exp(i\theta)$ and transmission $\tilde{t} = |\tilde{t}| \exp(i\theta)$ coefficients. They depend on the complex refractive index of the film $\tilde{n} = n - ik$ and the angle of incidence φ_0 . The reflectance, denoted R, is the ratio of reflected radiation intensity I_r and incident radiation intensity I_0 , and is the square of the complex reflection coefficient: $R = r^2 = |\tilde{r}|^2$. The transmittance, denoted T, is the ratio of transmitted radiation intensity I_t and incident radiation intensity I_0 and is the square of the complex reflection coefficient: $R = r^2 = |\tilde{r}|^2$.

To determine the optical constants of polymer films can be used both external specular reflection spectra and internal reflection spectra.

For the specular reflection on thin polymer films (with thicknesses less than $2\mu m$) deposited on metals, as in Figure 1, the reflected radiation contains two components: one reflected by air-film interface (I_r intensity) and one reflected by film-metal interface (I_t intensity), after having twice crossed the polymer film.

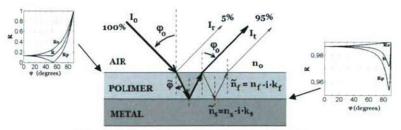


Fig.1 Specular reflection of radiation on thin films deposited on metals

The reflectance due to reflection at the air-film interface has values of about 5% because the values of refractive indices of polymers in the infrared region are less than 1.5. Meanwhile the reflectance at the polymer-metal interface has values over 99%. Thus, the radiation intensity reflected at the polymer-metal interface and twice crossing the polymer film is about 95% of incident radiation intensity, and almost 20 times greater than the beam directly reflected at the air-polymer interface. Therefore, this spectrum is a reflection-absorption (or transflectance) spectrum, and is similar in terms of quality of transmission spectrum of the polymer film. This spectrum can be processed to obtain the optical constants **n** and **k** corresponding to the complex refractive index $\tilde{n} = n - ik$ [7].

The two optical constants can be obtained either from the reflectance measurements at two angles of incidence or using the whole range of reflectance spectrum values at one angle of incidence [5]. In the second case, Kramers-Kronig analysis of the recorded spectrum is used [10,11].

The dispersion analysis is another way to obtain the optical constants of thin films deposited on metal surfaces. It is more flexible than Kramers-Kronig analysis and can be applied to more complicated geometry of the spectrum.

Dispersion analysis is based on building an appropriate model for dielectric function and calculating the optical properties corresponding to this model. The best known is Drude-Lorentz model [6, 9, 12] which defines the electric permittivity:

$$\varepsilon(\nu) = \varepsilon_{\infty} + \sum_{j} \frac{\nu_{pj}^2}{\nu_{0j}^2 - \nu^2 - i\gamma_j \nu}$$
(1)

It describes the optical response of a set of harmonic (damped) oscillators. In this relation, ε_{∞} is so-called "high-frequency dielectric constant", which represents the contribution of all oscillators at very high frequencies (compared to frequency range under consideration). The parameters v_{pj} , v_{0j} and γ_j are the "plasma" frequency, the transverse frequency (eigenfrequency), and the line-width (scattering rate), respectively of the j-th Lorentz oscillator. For the proposed model, from permittivity, we can calculate all optical quantities such as reflectance R and transmittance T. The spectrum of these theoretically calculated quantities is compared with those experimentally determined. The model parameters are continuously adjusted to fit the theoretical values with experimentally measured data. In case of reflection-absorption spectra the theoretical transmintance and experimental transmitance spectra are compared. The fitting parameters process stops when the differences between theoretical and experimental spectra are minimal.

Suppose, we have a set of N experimental data points $\{x_i, y_i, \sigma_i\}$ (j = 1, ..., N) that we want to fit. Here, x_i is the light frequency, y_i is the data value, and σ_i is the data error bar. For a set of M internal parameters, the values $y = f(x, p_1, ..., P_M)$ are calculated based on the model. The so-called Levenberg-Marquardt algorithm is used to minimize the value:

$$\chi^{2} = \sum_{j}^{N} \left(\frac{y_{j} - f(x_{j}, p_{j} \dots p_{M})}{\sigma_{j}} \right)^{2} = \chi^{2}(p_{1}, \dots, p_{M})$$
(2)

Fitting process stops when the stopping criterion is met [8].

2. EXPERIMENTAL

Thin films of polyvinyl acetate (PVAc) were obtained by spraying a polymer solution on the surface of a steel metal sample. Low concentration of polymer in the solvent allowed us to obtain thin films with thickness less than $2\mu m$. After evaporation of the solvent the IR reflection-absorption spectra at 20^{0} incidence angle was recorded. The metal sample coated with polymer film was then heated to a temperature of 80^{0} C for 30 minutes to complete removal of the solvent. After the sample heat treatment we recorded IR reflection-absorption spectrum again. The metal surface used as substrate for the polymer film was obtained by grinding and polishing.

The IR reflection-absorption spectra were recorded using a specular reflectance device for UR-20 spectrograph.

The reflection-absorption spectra were processed using the program RefFIT [8] to obtain the optical constants of polymer film.

3. RESULTS AND DISCUSSION

The IR reflection-absorption spectra recorded at 20 degrees incidence angle are shown in Figure 2. The spectral range $500 \div 1800 \text{ cm}^{-1}$ containing the absorption band at 1732 cm $^{-1}$ corresponding to C = O stretching vibration is presented [3].

The reflectance R and hence the transmitance T values depend very least on the angle of incidence, as observed in Figure 1. Because of this, reflection-absorption spectrum at 20 degrees angle of incidence is very similar to the transmission spectrum for normal incidence.

Since the surface film thickness is less than $2\mu m$, interference fringes in spectra recorded are not present [6]. For the same reason the spectrum recorded is one of reflection-absorption (transflectance).

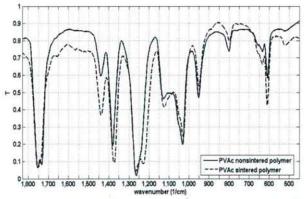


Fig.2 Reflection-absorption spectra (RA) at 20^o incidence angle for a PVAc film deposited on polished steel

The IR spectra of PVAc film deposited on steel recorded after its heat treatment, shows the appearance of a thermal degradation process of polymer, which is its depolymerization.

The depolymerization process produces monomer units containing unsaturated C = C bonds. This is confirmed by increased of the absorption band at 1647 cm⁻¹ characteristic for stretching frequency of C = C bond [2,3]. During thermal degradation is also possible to remove the acetic acid. This is highlighted by the change of vibration frequencies of C = O, C-O and C-C-O bonds. Thus, instead of stretching vibration frequencies of the C=O bonds at 1724 cm⁻¹ and 1737 cm⁻¹ there is a single band at 1750 cm⁻¹. Instead of the absorption band at 1241 cm⁻¹, corresponding to vibration frequency of C-O bonds in the acetate group linked to the macromolecular chain or partially released. Also, there is a decrease in the intensity of the absorption band at 1124 cm⁻¹ corresponding to the vibrations of C-C-O bonds [3].

The spectra obtained can be processed to obtain the optical constants \mathbf{n} (refractive index) and \mathbf{k} (absorption index).

We used 1381 points and 55 parameters in the fitting process.

To improve the accuracy of the dielectric function we used simultaneously the fitting process of reflection-absorption spectra recorded at a 20 degrees incidence angle and the refractive index of PVAc value $n_f = 1.4665$ [1].

The refractive index spectrum obtained by dispersion analysis of reflection-absorption spectrum for PVAc deposited on steel is shown in Figure 3.

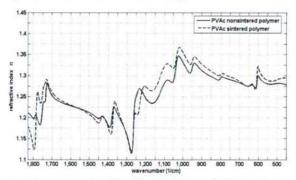


Fig. 3 Refractive index spectra for PVAc deposited on steel obtained by dispersion analysis

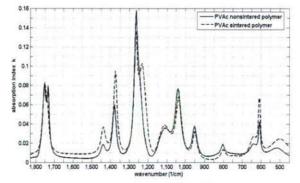


Fig. 4 Absorption index spectra for PVAc deposited on steel obtained by dispersion analysis

The absorption index (extinction coefficient) \mathbf{k} has a spectrum very similar to the absorption spectrum. The absorption coefficient spectrum of PVAc film deposited on steel is shown in Figure 4. The values of the absorption coefficient are low. The polymer can be considered to

be almost transparent in infrared. In this spectrum, the absorption bands do not show deviations from the positions of the absorption bands in reflection-absorption spectrum.

4. CONCLUSIONS

The IR reflectance spectra for thin films (with thickness less than $2\mu m$) are reflectionabsorption spectra. The recorded size is the transmitance. Appearance of the reflectionabsorption spectra is very similar in terms of quality to the transmission spectrum of the polymer.

The dispersion analysis of reflection-absorption spectra is more accurate and easier than Kramers-Kronig analysis. The Kramers-Kronig analysis is used especially for thick surface films or for bulk materials.

Simultaneously fitting for several types of information about the film surface leads to accurate values of optical constants \mathbf{n} and \mathbf{k} .

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