

## DIRECT PHOTOLYSIS OF FUMONISIN B<sub>1</sub> IN AQUEOUS MEDIUM

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### Abstract

Fumonisin (FBs), secondary metabolites produced by fungi of the genus *Fusarium* [1], represent a major threat as potential organic pollutants. The FBs have been detected in different types of water [2], as well as produced them by fungi in untreated surface water [3]. Advanced Oxidation Processes (AOPs), have a high potential for water purification, including the removal of hazardous substances and pathogens from different types of water. AOPs are based on physicochemical processes that produce mainly hydroxyl radicals ( $\bullet\text{OH}$ ), representing primary oxidants, which can lead to complete mineralization of pollutants. These processes can be initiated by UV or solar radiation. The photocatalytic degradation has become a powerful method for degradation and transformation of aflatoxin B<sub>1</sub> [4], zearalenone [5], and deoxynivalenol [6] into harmless substances. In this paper, we have investigated optimization of high performance liquid chromatography with fluorescence detector method for monitoring the stability of fumonisins B<sub>1</sub> (FB<sub>1</sub>), B<sub>2</sub> (FB<sub>2</sub>), and B<sub>3</sub> (FB<sub>3</sub>) solutions as well as the efficiency of FB<sub>1</sub> degradation using direct photolysis under UV and solar radiation in ultrapure water. It was found that the sensitivity and separation of the FB<sub>1</sub> peak from *o*-phthalialdehyde–2-mercaptoethanol (used for derivatization) was optimally at isocratic elution using the MeOH–NaH<sub>2</sub>PO<sub>4</sub> mobile phase, at a ratio of 75 : 25 (v/v). When studying the efficiency of direct photolysis of  $1.47 \times 10^{-6}$  mol/dm<sup>3</sup> solution of FB<sub>1</sub> it was found that after 180 min of irradiation degradation efficiency was 88% using UV and 76% using solar radiation at pH 8.2. Also, the effect of pH in the range from 4.0 to 10.0 on the efficiency of direct photolysis of FB<sub>1</sub> was examined.

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