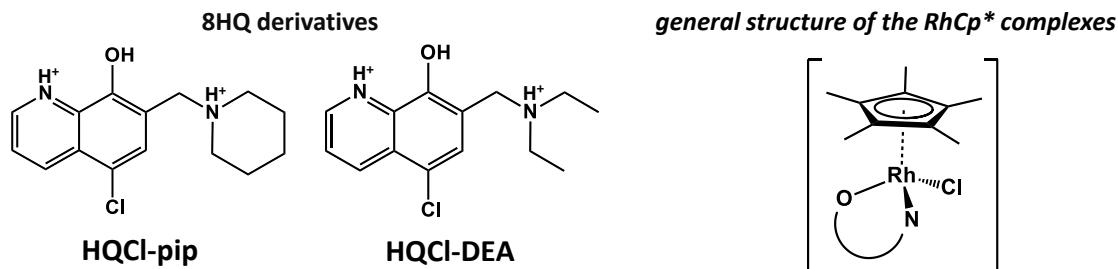


**A MULTIDRUG RESISTANCE SELECTIVE ANTICANCER  
8-HYDROXYQUINOLINE DERIVATIVE AND ITS RhCp\* COMPLEX**

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In the field of developing more effective chemotherapeutic drugs with fewer severe side effects, 8-hydroxyquinoline (8HQ) derivatives are promising compounds. A group of these molecules was found to be selective against multidrug resistant (MDR) cancer cells, which occurs frequently when treating cancer [1]. Furthermore, several studies showed that these molecules form complexes with essential metal ions which is associated with the therapeutic effect [2]. In our research group several 8HQ Mannich base compounds have been developed and tested against MDR cancer cells, including HQCl-pip. Its complexation with half-sandwich  $\text{Rh(III)(η}^5\text{-petamethylcyclopentadienyl)(H}_2\text{O)}_3\text{]}}^{2+}$  ( $\text{RhCp}^*$ ) cation (see figure), has actually enhanced, rather than diminished the toxicity on MDR cells; moreover, the complex has better aqueous solubility in comparison to HQCl-pip alone [3].



HQCl-DEA is a close derivative of the former compound showing selective toxicity on MDR cancer cells. Therefore, we found it worth to investigate its behaviour in aqueous solution, complex formation with  $\text{RhCp}^*$  cation and their interaction with human serum albumin. Proton dissociation processes and stability of HQCl-DEA and its  $\text{RhCp}^*$  complex was followed by UV-Vis spectrophotometry and  $^1\text{H}$  NMR spectroscopy. The same techniques were applied to determine the  $\text{Cl}^-/\text{H}_2\text{O}$  exchange constant of the complex. Lipophilicity was determined by *n*-octanol/water partitioning experiments at various pH values and chloride ion concentrations. Albumin binding was investigated by equilibrium dialysis, ultrafiltration, UV-Vis and fluorescence spectroscopies. Based on our results, the complex shows promise as an anticancer agent against MDR.

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